

Electricity -

Electricity can travel through some materials better than others

GOOD CONDUCTORS:

POOR CONDUCTORS:

Modern materials designed to conduct only under certain conditions.

The ability of a material to conduct electricity (for electrons to flow through the material) depends on the atomic structure.

29
Cu
63.546

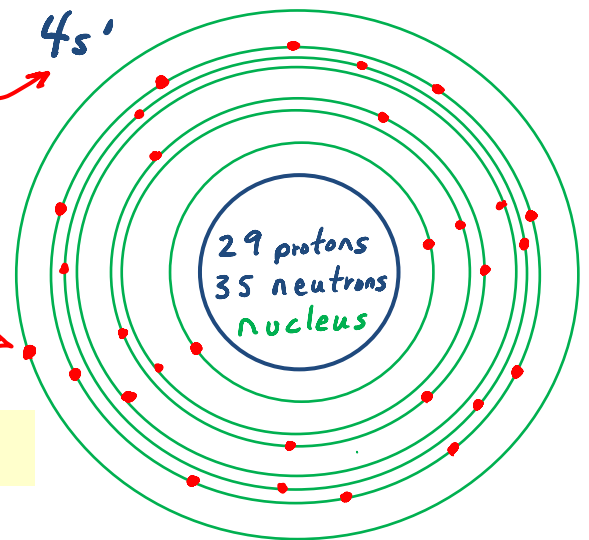
← Atomic number of copper =
29 protons + 29 electrons
← Atomic weight () + 35 neutrons
64
← # protons + avg. # neutrons = atomic wt.

Lightweight electrons whiz about the nucleus (we think of them as being in shells)



electrons in outer shell influence conductivity

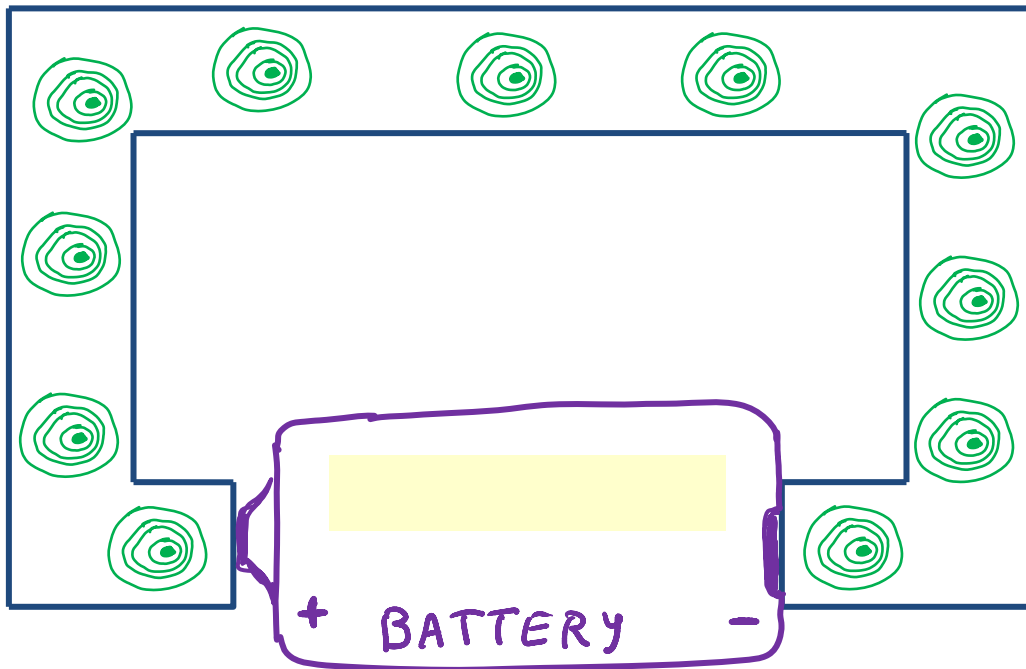
It is EASY for copper to let the 4s¹ electron go, so copper is a GOOD



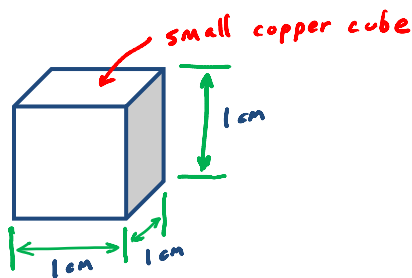
hold onto their electrons more tightly due to their electron configuration.

If you were talking with someone in China on the phone, your voice would cause the diaphragm in the microphone to move which would induce electron motion. Would an electron that you just caused to move in the microphone somehow get to China in a fraction of a second? (assuming a telephone line was stretched from here to China)

Flow of electricity through a copper wire:



How many electrons are we talking about?



Atomic wt. of Cu = $63.55 \frac{\text{g}}{\text{mol}}$

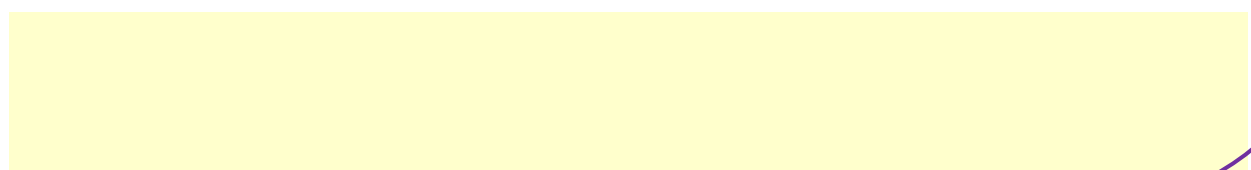
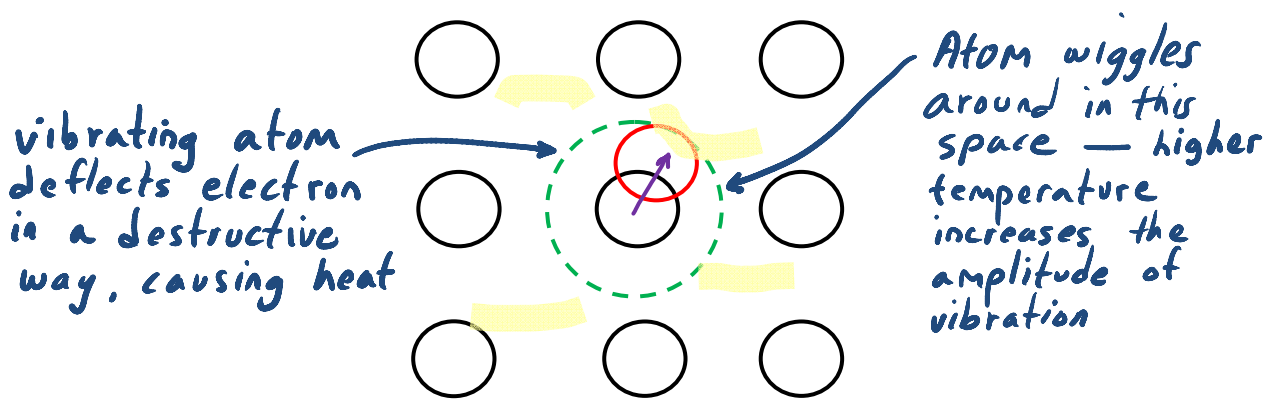
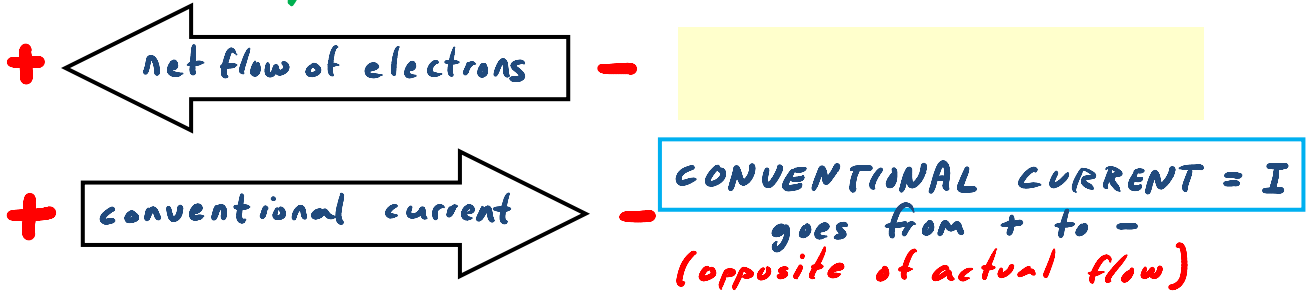
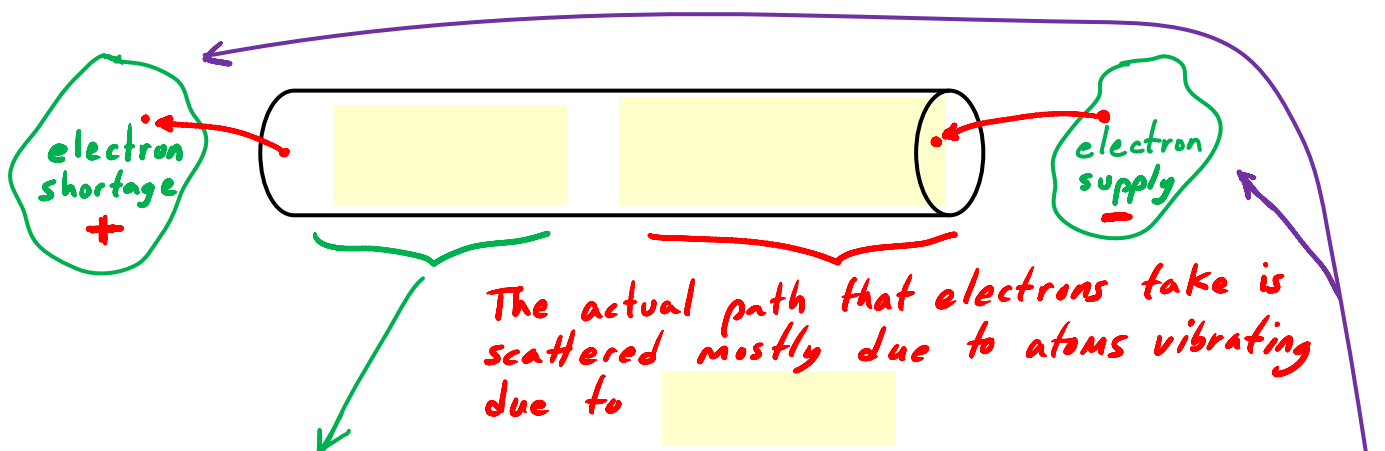
Density of Cu = $8.94 \frac{\text{g}}{\text{cm}^3}$

Avogadro's Number = $6.022(10)^{23} \frac{\text{atoms}}{\text{mol}}$

atoms = # valence electrons for Cu =

valence electrons =

↳ This is more than the number of grains of sand on the entire earth (assuming grains of sand 1mm in diameter, packed at 68% efficiency, 10cm deep over the earth's 200 million square mile surface)



DEFINITIONS

Current = $I =$

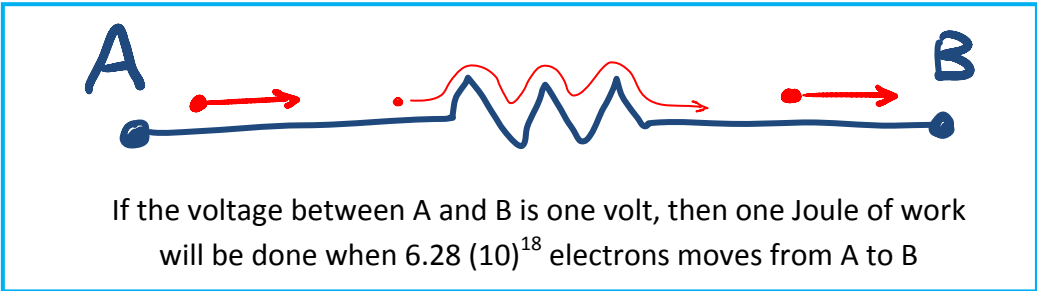
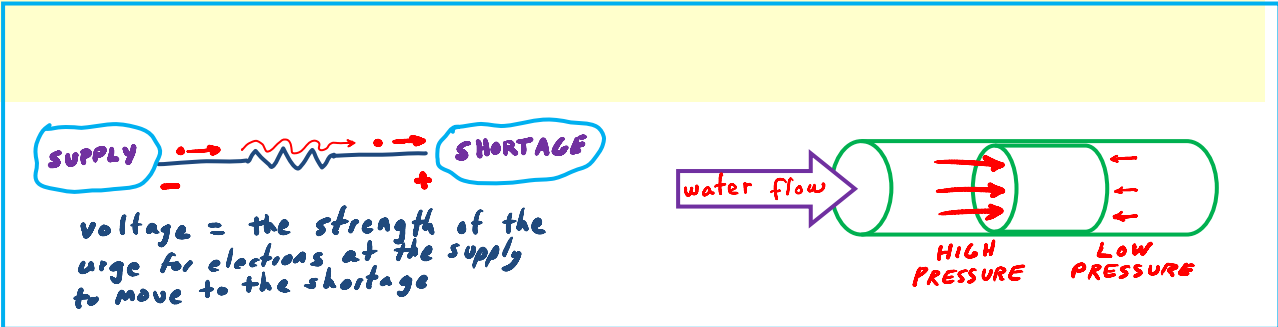
$$1 \text{ Amp} = \frac{6.28(10)^{18} \text{ electrons passing a point}}{\text{per second}} = 1 \frac{\text{Coulomb}}{\text{s}}$$

$$1 \text{ C} = 6.28(10)^{18} \text{ electrons}$$

teams of 2

CLASS PROBLEM: A battery powers a flashlight. If the battery supplies a steady current of 1.3 A over 8 minutes, how many electrons leave the negative terminal of the battery during this time period?

Analogy between electric voltage and water pressure



Resistance = R

$$R = \text{ohms or } \Omega$$

Ohm's Law

CLASS PROBLEM: A 1.5 volt AA battery is wired to a light bulb that has a resistance of $30\ \Omega$.

- (a) Draw the circuit
- (b) Find the current

teams of 2