

RADIATION PHYSICS

① KIRCHOFF'S LAW

• Any body $T > 0\text{ K}$ radiates energy

• Radiation proportional to absorption

W_i = power density of radiation (power/area)
= energy FLUX (energy/time/area.)

a_i = specific absorptivity ($0 < a_i < 1$)

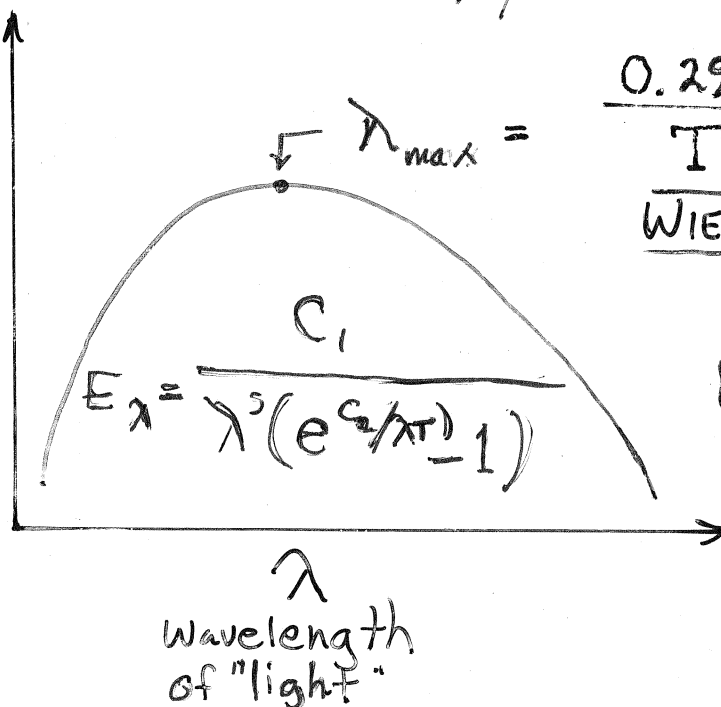
$$\frac{W_1}{a_1} = \frac{W_2}{a_2} = W_B = \text{black body radiator power density}$$

② PLANCK'S LAW

Radiation from a black body is distributed w/r/t wavelength

$$W_B = \int_0^{\infty} W_B^{\lambda} d\lambda$$

log relative emission



$$0.29 \text{ cm}\cdot\text{K}$$

T

WIEN DISPLACEMENT LAW

E = total emissive power

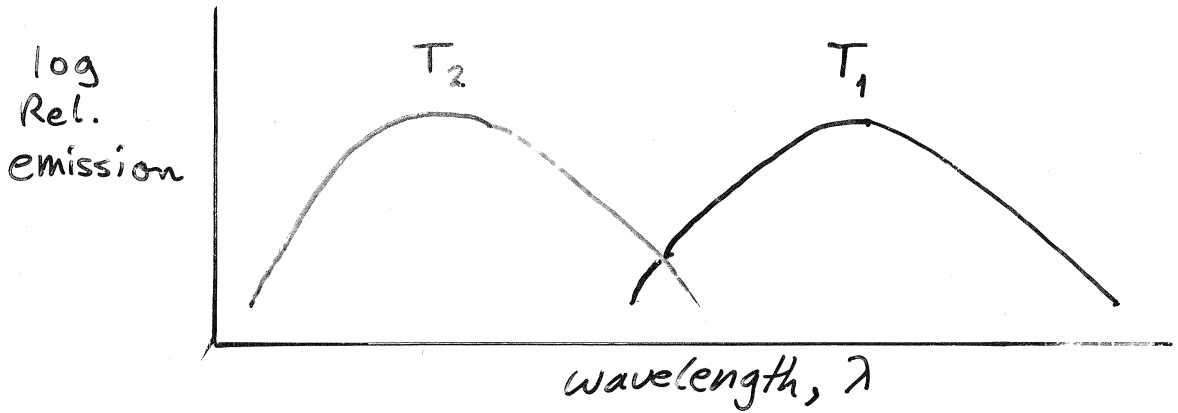
$$= W_B \cdot A$$

3

STEFAN-BOLTZMANN
LAW

CONSIDER TWO BLACK BODIES

$$T_2 \gg T_1$$



Which body is radiating more thermal energy?

BLACK BODY:

$$W_B = \sigma T^4$$

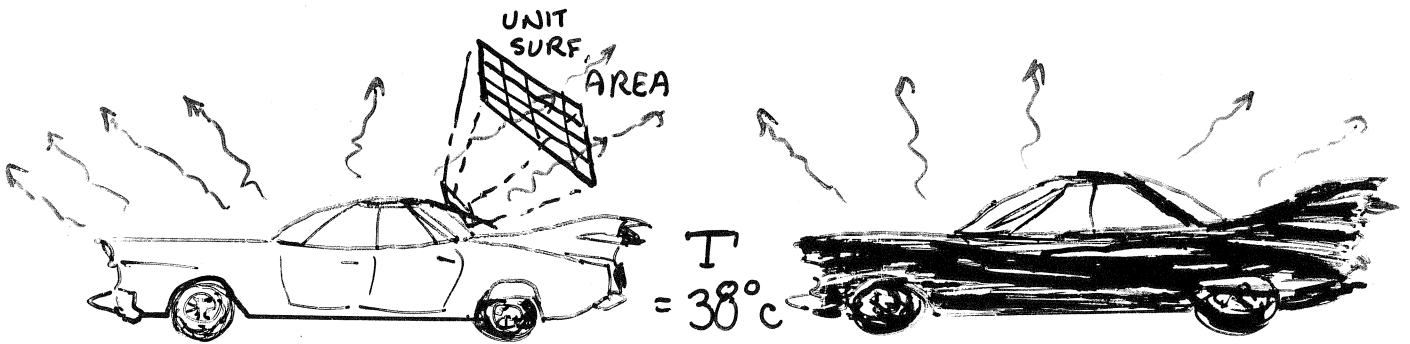
$$\sigma = 8.3 \times 10^{-11} \text{ cal cm}^{-2} \text{ min}^{-1} \text{ K}^{-4}$$

$$= \underline{5.7} \times 10^{-8} \text{ J m}^{-2} \text{ s}^{-1} \text{ K}^{-4}$$

ANY REAL BODY:

$$W_i = \epsilon_i \sigma T^4$$

ϵ_i = emissivity = a_i at thermal equilibrium



$$W_1 = (0.3)(8.35 \times 10^{-11})(311)^4$$

$$= 0.23 \text{ cal cm}^{-2} \text{ min}^{-1} \left(\frac{\text{Langley}}{\text{min}} \right)$$

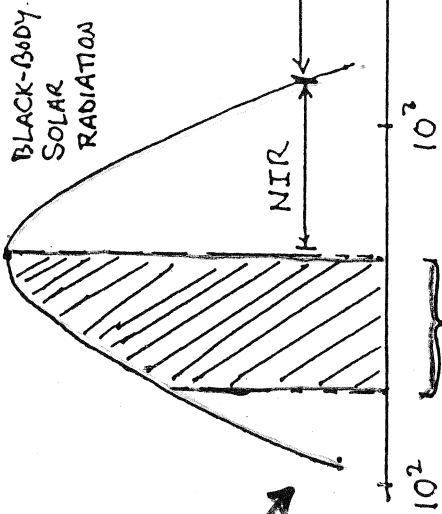
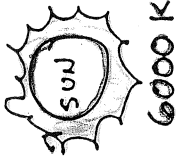
$$= 160 \text{ Watt/m}^2$$

$$W_2 = (0.8)(8.35 \times 10^{-11})(311)^4$$

$$= 0.62 \text{ langley/min}$$

$$= 434 \text{ W/m}^2$$

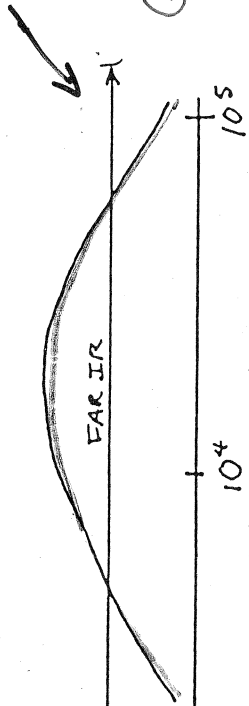
93 million miles



BLACK-BODY TERRESTRIAL RADIATION

$W_B^{EARTH} = 0.56 \text{ ly min}^{-1}$

Ⓒ EARTH'S SURFACE



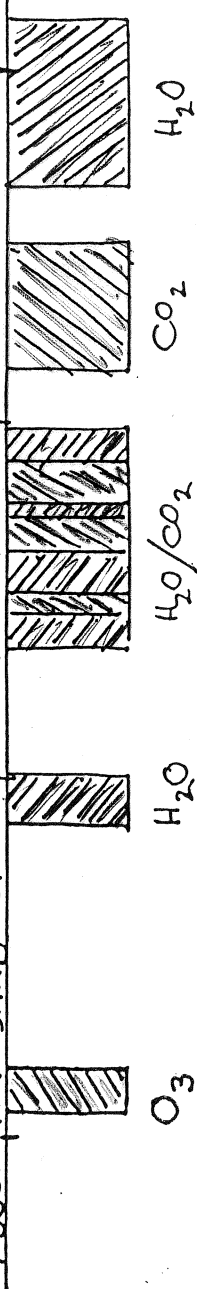
$W_B^{SUN} = 107,000 \text{ ly min}^{-1}$
(AT SURFACE OF SUN)

WAVELENGTH (nm)

400-800 nm

VISIBLE LIGHT

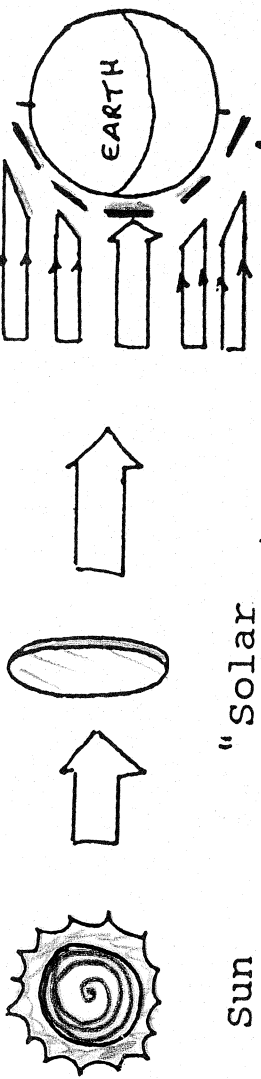
ABSORPTION BANDS OF THE EARTH'S ATMOSPHERE



ATMOSPHERE IS TRANSPARENT TO MOST INCOMING SOLAR RADIATION (Visible and near-IR)

ATMOSPHERE ABSORBS HEAVILY THE OUTGOING FAR-IR RADIATION OF THE EARTH
"GREENHOUSE EFFECT"

The spherical shape of the earth is very important:



At higher latitudes, light is spread over greater area; hence, less intense.

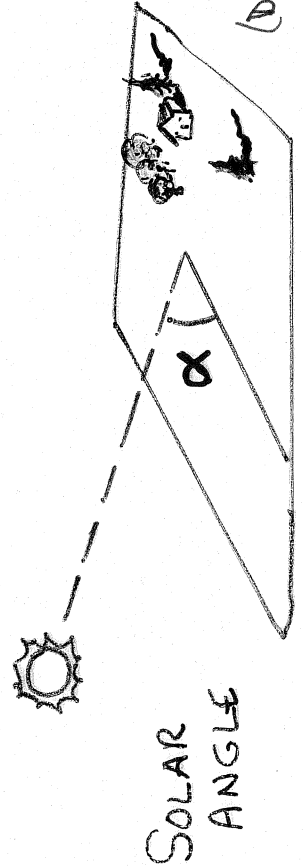
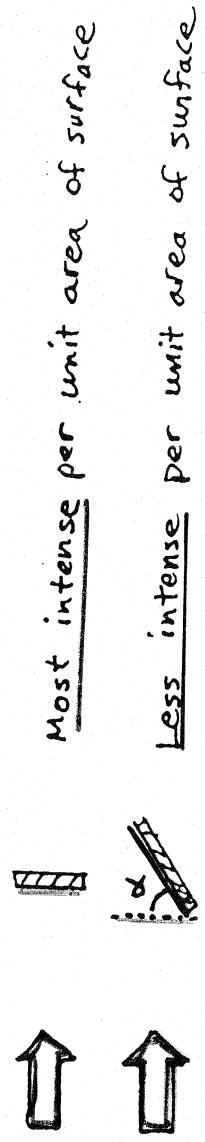
"Solar constant"

$W_B^0 \approx 2.0$ langley/min $\approx 1,400$ W/m² (IN OUTER SPACE)
 $(\text{cal/cm}^2/\text{min})$
 $< 0.001\%$ OF W_B AT SURFACE OF SUN

"Insolation" = I_0

$I_0 = W_{B0} \sin \alpha$
 $= W_{B0} \cos \phi$
 $\alpha = [90^\circ - (\text{latitude})]$ for noon at the equinox

α = solar angle (See Eagleson's chapter)



$$\sin \alpha = \sin \delta \sin \Phi + \cos \delta \cos \Phi \cos \tau$$

- (DELTA) δ = DECLINATION
- (PHI) Φ = LOCAL LATITUDE
- (TAU) τ = HOUR ANGLE

SOLAR ANGLE CALCULATION

$$\sin \alpha = \sin \delta \sin \Phi + \cos \delta \cos \Phi \cos \tau$$

δ = declination

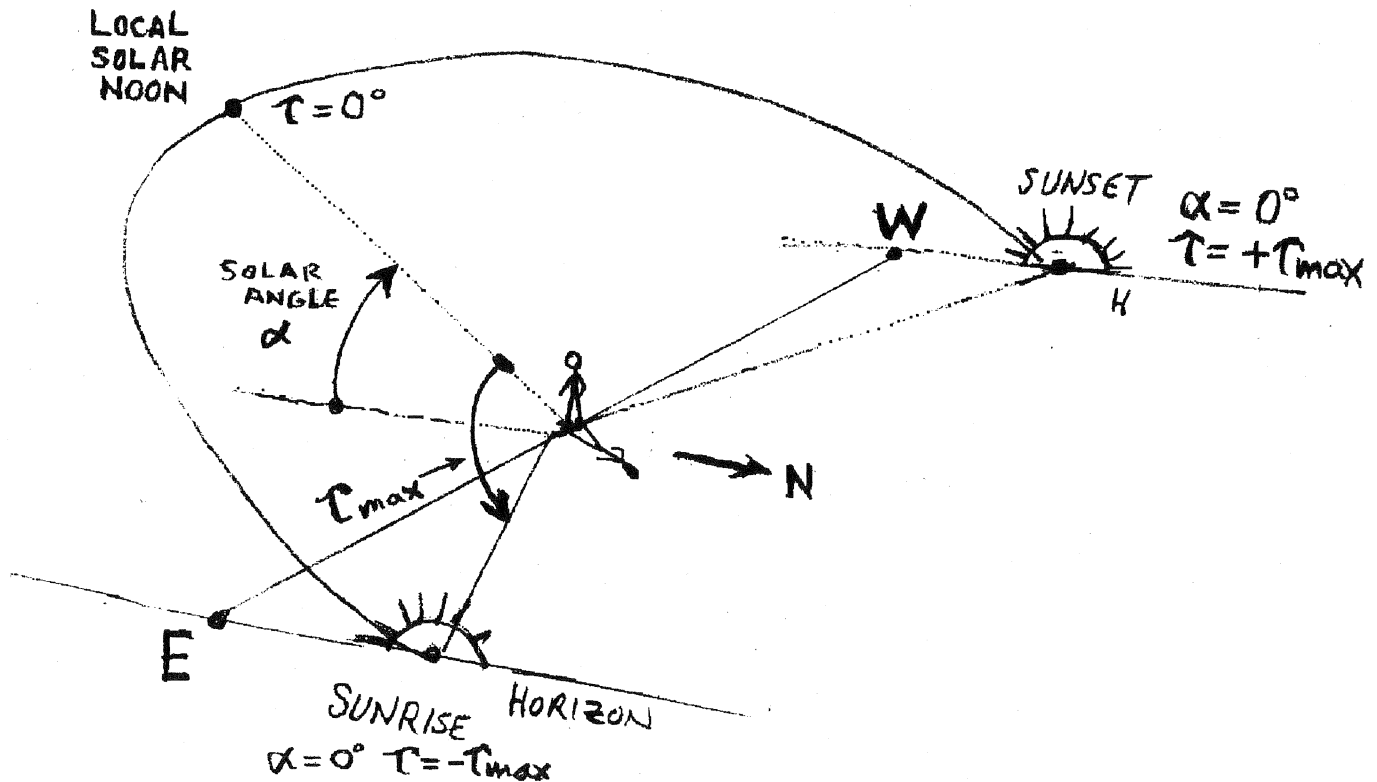
(the seasonally varying angle of the plane traversed by the sun across the sky)

Φ = latitude

(the geographic angle along the surface of the earth)

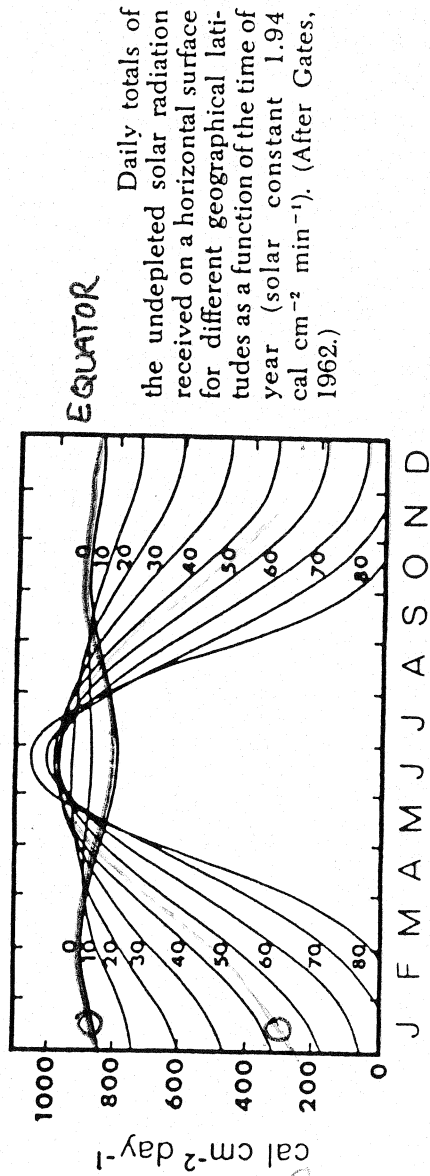
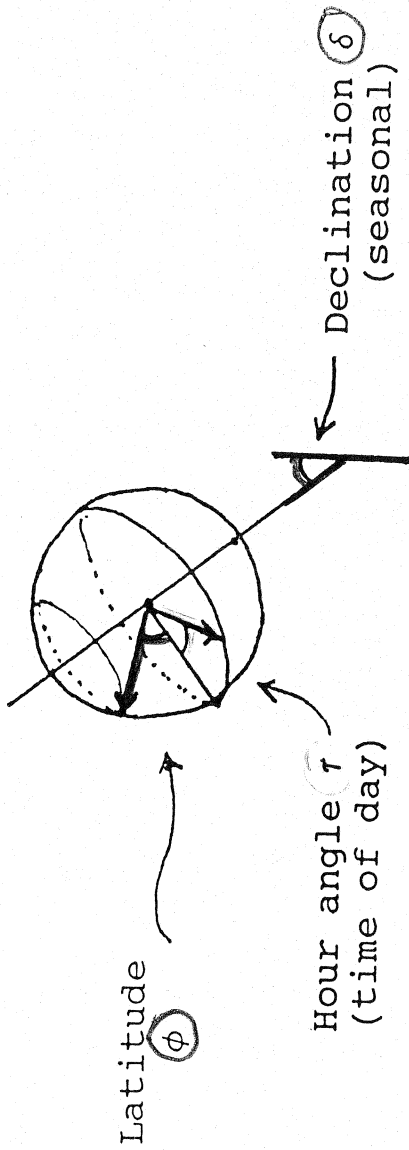
τ = hour angle of the sun

(the angle of the sun *along* the arc traversed by the sun across the sky)



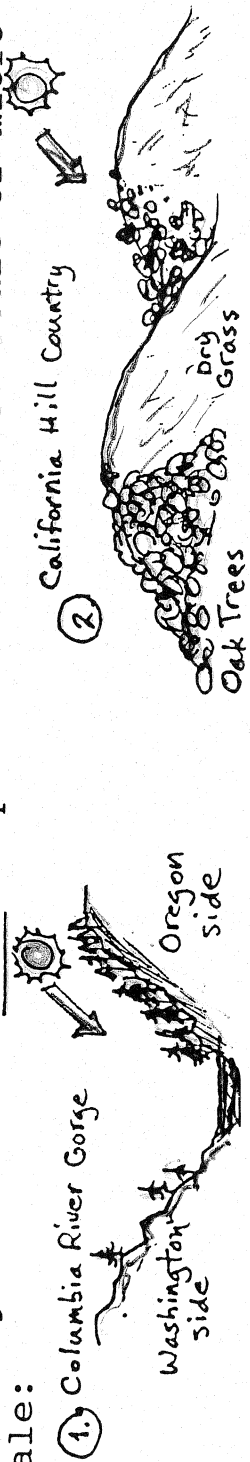
A.

Solar angle for a flat surface can be calculated using the three component angles: ϕ , τ , δ .

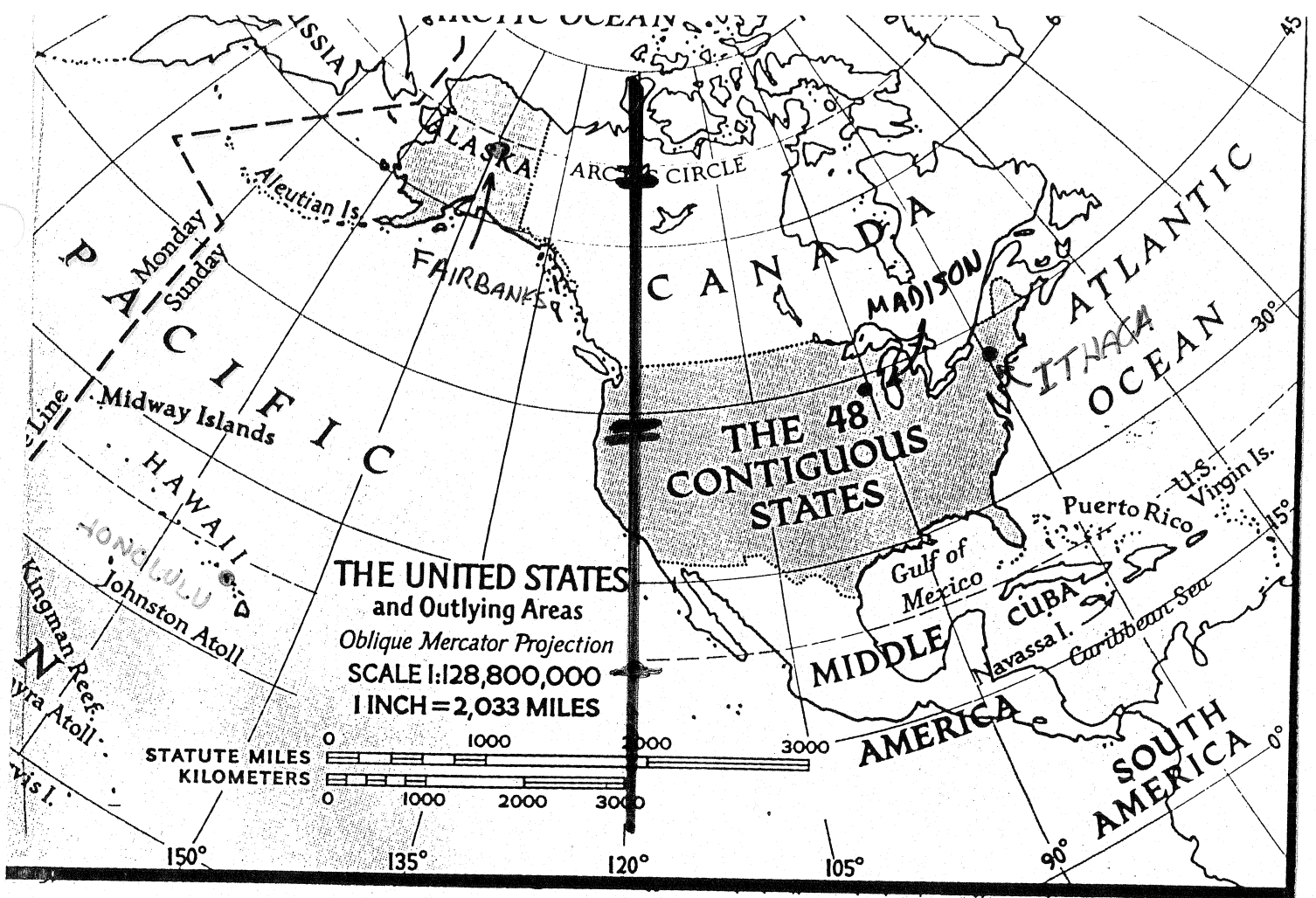


B.

Solar angle also has local components on the meso-scale or micro-scale:



Absorbance and especially molecular and particulate scattering also affect insolation: smoke, haze, clouds, pollution, etc. So what hits the surface is significantly less than ($W_{Bo} \sin \alpha$).



COMPARATIVE LATITUDES OF FOUR U.S. CITIES

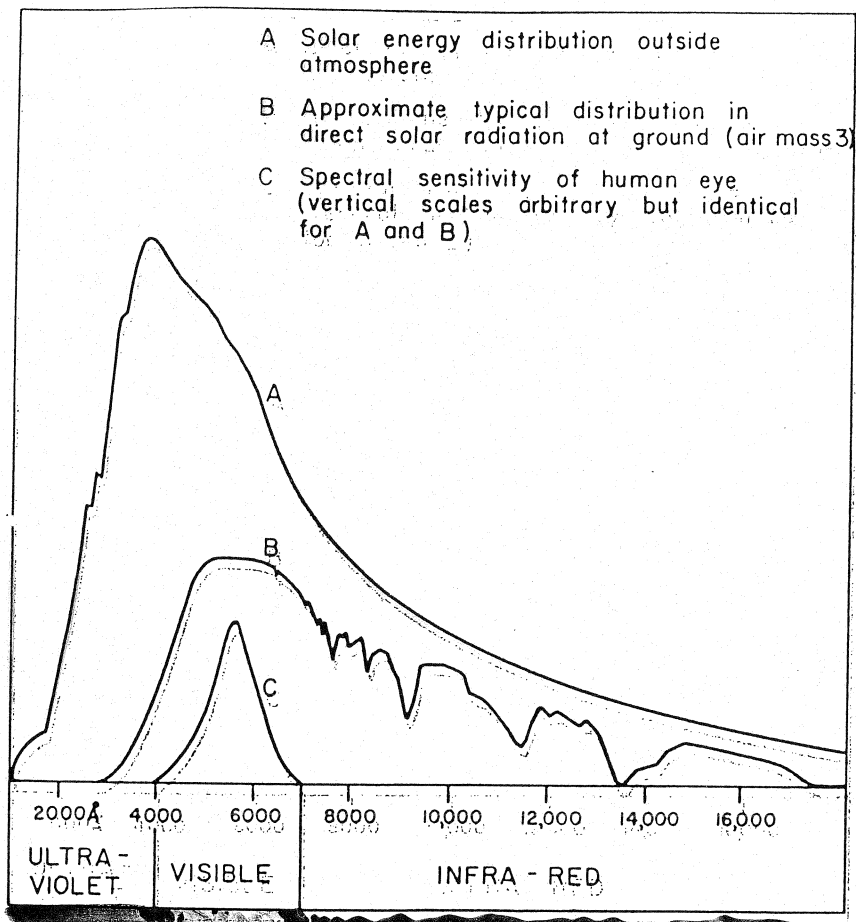


FIGURE 109. Spectral distribution of radiant energy outside the atmosphere and at the earth's surface (air mass 3), with the spectral sensitivity of the human eye.

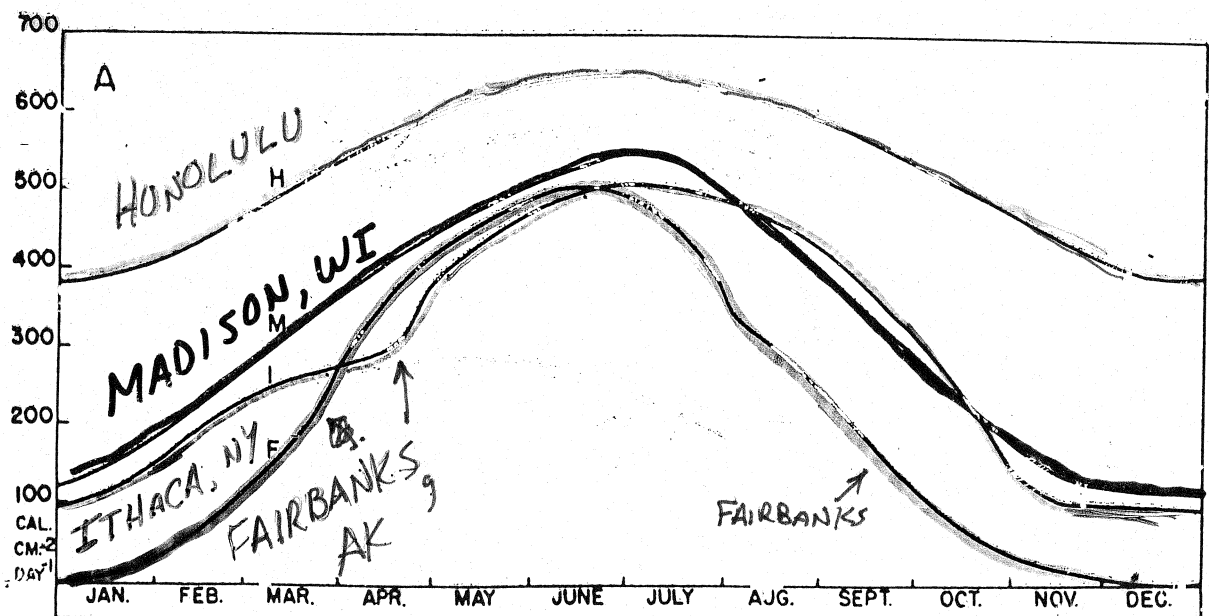


FIGURE 111. A, mean radiation (cal. cm.⁻² day⁻¹) received at various times of year at H, Honolulu, Hawaii, lat. 21°18' N.; M, Madison, Wisconsin, lat. 43°05' N.; I, Ithaca, New York, lat. 42°27' N.; F, Fairbanks, Alaska, lat. 64°52' N. (after Hand). Note the slightly lower radiation received at nearly all seasons at Ithaca (Lake Cayuga) as compared with Madison (Lake Mendota).

BEER-LAMBERT-BOUGER LAW INTEGRATE w/r/t DEPTH (z)

$$I_z^\lambda = I_0^\lambda e^{-a_\lambda z}$$

OR

a_λ = absorption coeff
for wavelength λ

$$\frac{I_z^\lambda}{I_0^\lambda} = e^{-a_\lambda z}$$

①

LIGHT INTENSITY
DROPS OFF
EXPONENTIALLY
w/ DEPTH (z)

② EXTENT OF LIGHT ATTENUATION
DEPENDS ON WAVELENGTH OF LIGHT

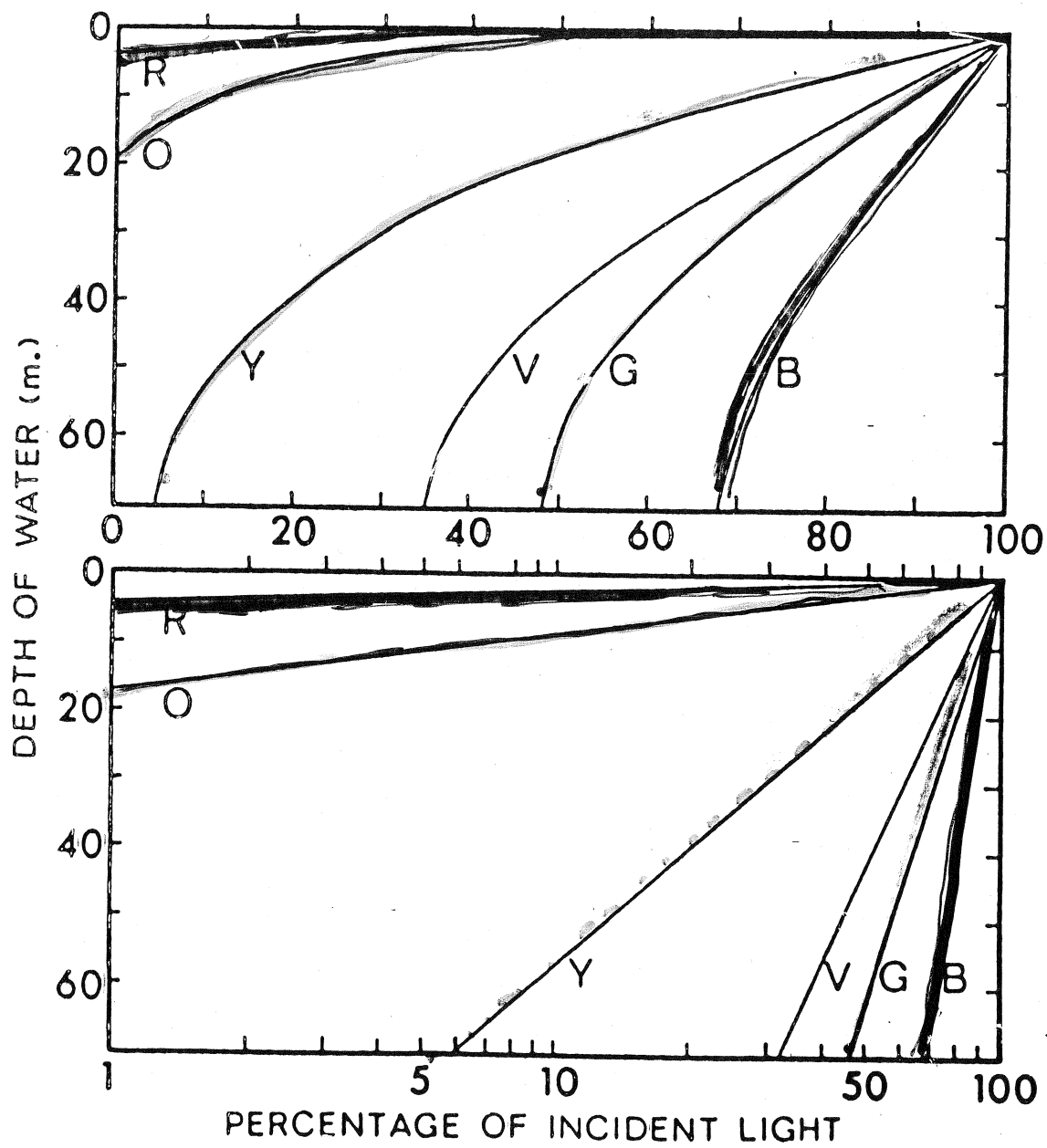
● RED LIGHT: $a_{650 \text{ nm}} \rightarrow$ LARGE

● GREEN LIGHT $a_{500} \rightarrow$ MEDIUM

● BLUE LIGHT $a_{450} \rightarrow$ SMALL

● UV LIGHT $a_{300} \rightarrow$ VERY LARGE

"a" also known as the "Extinction Coeff."

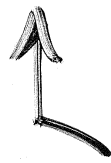


PRACTICAL APPLICATION

Beer's Law in Natural Water

DEFINE $K \equiv \int_a^\lambda a_\lambda d\lambda$

$K \equiv$ Average absorption of
sunlight for natural water



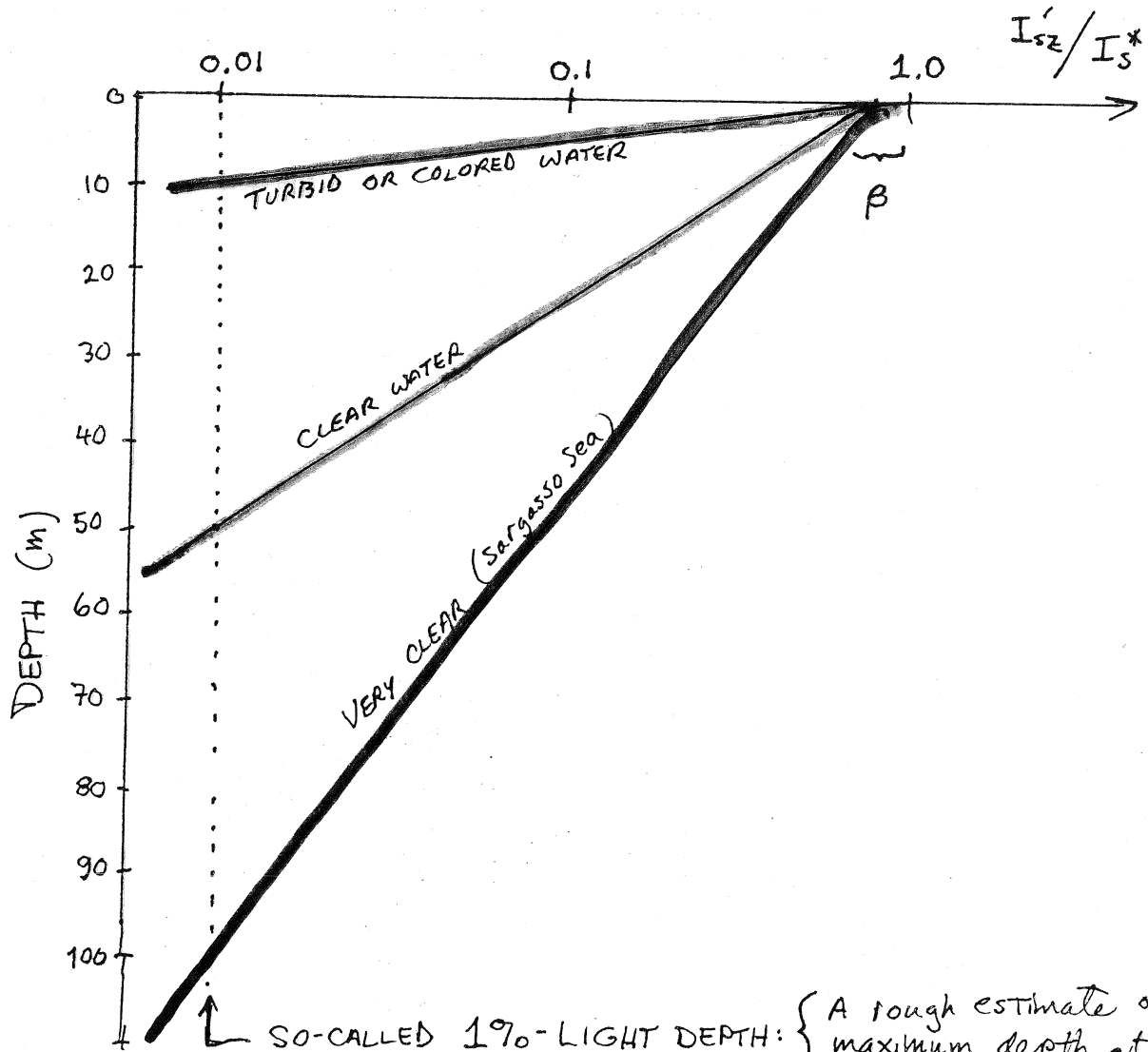
OFTEN MEASURED
EMPIRICALLY W/ LIGHT METER

$\beta \equiv$ "EXTRA" absorption often
seen in film on surface

$$\frac{I_{sz}}{I_s^x} = (1 - \beta) e^{-Kz}$$

ABSORPTION OF LIGHT BY WATER: Logarithmic Presentation

The exponential equation given by Beer's Law can be linearized by taking the natural log of both sides, or, by plotting light intensity on semi-log scaling.



SO-CALLED 1% LIGHT DEPTH: { A rough estimate of the maximum depth at which photosynthesis can occur.